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		TR	ANSMITTAL LETTER	U.S. APPLICATION NO. (If known, see 37 CFR 1.5)					
]	DESIGNATED/ELECTE	ED OFFICE (DO/EO/US)	00/00071/				
				G UNDER 36 U.S.C. 371	09/980346				
ļ	INT	ERN	ATIONAL APPLICATION NO.	PRIORITY DATE CLAIMED					
	PC]	/SE	00/01152	02 June 2000 (02.06.00)	03 June 1999 (03.06.99)				
		TITLE OF INVENTION							
L			ING DEVICE						
			ANT(S) FOR DO/EO/US						
ŀ		Gunnar Skarping, Marianne Dalene Applicant herewith submits to the United States Designated/Elected Office (DO/EO/US) the following items and other information:							
	[V]								
1	1.				5 II S C 371				
	2.			T submission of items concerning a filing under 3: examination procedures (35 U.S.C. 371(f)). The s					
	3.		This express request to begin national (21) indicated below.	examination procedures (33 0.3.C. 371(1)). The s	administration metado nomo (2), (0), (2) and				
	4.			ation of 19 months from the priority date (Article 3	1).				
	4 . 5.								
	٥.	A copy of the International Application as filed (35 U.S.C. 371(c)(2)) a. is attached hereto (required only if not transmitted by the International Bureau).							
		 a. is attached hereto (required only if not transmitted by the International Bureau). b. has been communicated by the International Bureau. 							
			c. is not required, as the application was filed in the United States Receiving Office (RO/US).						
	6.		An English language translation of the	a language translation of the International Application as filed (35 U.S.C. 371(c)(3)).					
1			a. is attached hereto.						
State of the first			had been previously submitted under 35 U.S.C. 154(d)(4).						
	7.	X	Amendments to the claims of the Inter	rnational Application under PCT Article 19 (35 U.	S.C. 371(c)(3))				
			a. are attached herewith (required only if not transmitted by the International Bureau).						
in Head			b. have been communicated by the International Bureau.						
Ļj	ور		c. have not been made; however, the time limit for making such amendments has NOT expired.						
		_	d. A have not been made and will						
113	8: An English language translation of the amendments to the claims under PCT Article 19 (35 U.S.C. 371(c)(3)).								
	9. An oath or declaration of the inventor(s) (35 U.S.C. 371(c)(4)). 10. A translation of the annexes to the International Preliminary Examination Report under PCT Article 36 (35 U.S.C. 371(c)(5)).								
		Iten	ns 11. to 20. below concern documen						
	11.		An Information Disclosure Statement						
	12. An assignment document for recording. A separate cover sheet in compliance with 37 CFR 3.28 and 3.31 is included.								
	13. X A FIRST preliminary amendment.								
	14.		A SECOND or SUBSEQUENT preli	minary amendment.					
	15.		A substitute specification.						
	16. A change of power of attorney and/or address letter.								
	17.		-	uence listing in accordance with PCT Rule 13ter.2	and 35 U.S.C. 1.821-1.825				
	18. A second copy of the published international application under 35 U.S.C. 154(d)(4).								
	19.			ge translation of the international application unde	r 35 U.S.C. 154(d)(4).				
20. Other items or information:									
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	Pag	ge 1	of 2		EV 000050790 HG				

	U.S. APPI	ICATION NO. (If known		INTERNATIO	NAL APPLICATION NO)	ATTORNEY'S DOO	CKET NUMBER
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	21. The following fees are submitted: BASIC NATIONAL FEE (37 CFR 1.492 (a) (1) - (5)):					CALCULATIONS	PTO USE ONLY	
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	and	International Search	n Report not prep	bared by the	EPO of JPO	\$1,040.00		
	International preliminary examination fee (37 CFR 1.482) not paid to USPTO but International Search Report prepared by the EPO or JPO \$890.00							
		mational preliminar international search				SPTO \$740.00		
	but a	mational preliminar all claims did not sa	tisfy provisions	of PCT Artic	cle 33(1)-(4)	\$710.00		
		mational preliminar all claims satisfied	provisions of PC	T Article 33	(1)-(4)	\$100.00		
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IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re application : GUNNAR SKARPING, ET AL.

Application No.

Filed : Herewith

For : SAMPLING DEVICE

Examiner

Attorney's Docket : AWA-043XX

Group Art Unit:

I hereby certify that this correspondence is being deposited with the United States Postal Service as first class mail in an envelope addressed to: Commissioner for Patents, Washington, D.C. 20231 on

Charles L. Gagnebin III
Registration No. 25,467
Attorney for Applicant(s)

PRELIMINARY AMENDMENT

BOX PCT Commissioner for Patents Washington, D.C. 20231

Sir:

Kindly enter the following Preliminary Amendment in the above-identified application:

In the Claims:

Please amend all the Claims to read as follows (a copy of the amended claims showing the additions and deletions appears at the end for the Examiner's convenience):

Express Mail Number

EV 009950780 US

Filed: Herewith

Group Art Unit:

1. A sampling device for analysis of a substance, which is

consisting of selected from the group

aminoisocyanates, isothiocyanates, amines and carboxylic acids

and which is present in both gas and particle phase in an air

flow intended to pass through the sampling device, characterized

in that it comprises

an adsorption device intended for the passage of the air

flow and provided with a coating of a mixture of a reagent in

the form of primary or secondary amines and a carboxylic acid

for adsorption of and reaction with the substance in the gas

phase of the air flow,

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a filter device intended for the passage of the air flow b)

and provided with the mixture of reagent and carboxylic acid for

adsorption of and reaction with the substance in the particle

phase of the air flow, and

a reagent container containing the reagent,

the reagent container being connected to the adsorption

device and/or the filter device by means of a switch device for

conveying the reagent to these for reaction therein with the

non-reacted substance.

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Filed: Herewith

Group Art Unit:

2. A sampling device according to claim 1, characterized in

that the adsorption device is tubular in the air flow direction,

the proportion of the length to the inner diameter being

superior to 5, and preferably about 10.

3. A sampling device according to claim 2, characterized in

that the inner walls of the adsorption device are coated with

the mixture of reagent and carboxylic acid.

4. A sampling device according to claim 1, characterized in

that the mixture contains one or more different carboxylic

acids, preferably formic acid, acetic acid or propionic acid,

or a combination thereof.

5. A sampling device according to claim 1, characterized in

that the reagent is di-n-butylamine (DBA).

6. A sampling device according to claim 1, characterized in

that the primary or secondary amines are volatile in unbound

form and each have a molecular weight inferior to 300.

7. A sampling device according to claim 1, characterized in

that the adsorption device has an upper end with an air inlet

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Filed: Herewith

Group Art Unit:

and a lower end which is connected to an upper end of the filter

device, the filter device having a lower end with an air outlet.

8. A sampling device according to claim 7, characterized in

that the sampling device is inverted in such manner that the air

inlet is arranged in the lower end of the filter device and that

the air outlet is arranged in the upper end of the adsorption

device.

9. A sampling device according to claim 1, characterized in

that it is provided with a pump or suction device to provide the

passage of the air flow through the adsorption device and the

filter device, and that it is connected either to the adsorption

device or to the filter device.

10. A sampling device according to claim 9, characterized in

that the pump or suction device is connected to the lower end of

the filter device.

11. A sampling device according to claim 10, characterized

in that the pump or suction device is a vacuum tube or a

displacement pump, preferably a hose pump, diaphragm pump,

injection pump or gear-type pump.

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Filed: Herewith

Group Art Unit:

12. A sampling device according to claim 1, characterized in

that the adsorption device consists of a body packed with

particles which are coated with the mixture of reagent and

carboxylic acid.

13. A sampling device according to claim 1, characterized in

that the filter device is formed as a substantially flat

cylinder having a diameter which is greater than or equal to

that of the adsorption device, that it contains particles which

are coated with the reagent and mixed with carboxylic acid, and

that it has an average pore diameter of 0.1-20 μm , preferably

0.2-2 $\mu\text{m},$ and most preferably 0.4 $\mu\text{m}.$

14. A sampling device according to claim 1, characterized in

that the reagent container is connected by means of the switch

device to the adsorption device or the filter device for auto-

matic conveyance thereto of reagent after completed sampling and

closing of the air inlet and the air outlet, respectively, for

reaction with residual non-reacted substance in the sampling

device.

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Filed: Herewith

Group Art Unit:

15. A sampling device according to claim 2, characterized in

that the reagent container is arranged as a concentric tube

round the adsorption device.

16. A sampling device according to claim 1, characterized in

that the adsorption device and the filter device are enclosed in

a uniform casing having a closable air inlet and outlet, respec-

tively.

17. A kit comprising a plurality of sampling devices

according to claim 1 for taking samples from a plurality of the

different substances at the same time or at different points of

time, the reagent in each sampling device being selected for the

substance with which it is to react in the air flow.

18. A method for immobilization of volatile primary and

secondary amines, preferably di-n-butylamine, on a surface,

characterized in that the volatile primary or secondary amine is

mixed with a carboxylic acid, preferably formic acid,

acid or propionic acid, and that the surface is subsequently

coated with the prepared mixture.

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Filed: Herewith

Group Art Unit:

19. A method for immobilization of volatile primary and

secondary amines, preferably di-n-butylamine, on a surface,

characterized in that:

the volatile primary or secondary amine is mixed with a

acetic acid carboxylic acid, preferably formic acid,

propionic acid, and that the surface is subsequently coated with

the prepared mixture;

one or more different primary and/or secondary amines,

preferably di-n-butylamine, is/are immobilized in an adsorption

device and/or a filter device in a sampling device or a kit

according to claim 1.

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20. A method for taking samples from a substance, which is

consisting isocyanates, of selected from the group

aminoisocyanates, isothiocyanates, amines and carboxylic acids

and which is present in both gas and particle phase in an air

flow, by means of a sampling device according to claim 1, char-

acterized in that it comprises the steps of

passing the air flow through the adsorption device and the

filter device by means of the pump or suction device

adsorption and reaction therein of the substance in both gas

phase and particle phase,

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Filed: Herewith

Group Art Unit:

closing the inlet and the outlet of the sampling device

after the passage of a predetermined volume of air, the switch

device associated with the reagent container automatically

securing conveyance of the reagent therein to the adsorption

device and the filter device for reaction therein with the

residual, non-reacted substance, and of

finally subjecting the sampling device to a qualitative

and/or quantitative analysis,

optionally carrying out an initial activation step, in which

the mixture of the reagent and the carboxylic acid is added to

the adsorption device and the filter device immediately before

the initiation of the sampling.

21. A method according to claim 15, characterized in that

aminoisocyanates, isothiocyantes, amines and isocyanates,

carboxylic acids are adsorbed by and react with di-n-butylamine

in the sampling device.

22. A method according to claim 16, characterized in that an

air flow of 0.001-1 l/min, and preferably 10-100 ml/min, is

passed through the sampling device.

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Filed: Herewith

Group Art Unit:

Please add the following new claims 23-27:

A sampling device according to claim 2, characterized 23.

in that the mixture contains one or more different carboxylic

acids, preferably formic acid, acetic acid or propionic acid,

or a combination thereof.

A sampling device according to claim 3, characterized

in that the mixture contains one or more different carboxylic

acids, preferably formic acid, acetic acid or propionic acid,

or a combination thereof.

A sampling device according to claim 2, characterized 25.

in that the reagent is di-n-butylamine (DBA).

A sampling device according to claim 3, characterized

in that the reagent is di-n-butylamine (DBA).

A sampling device according to claim 4, characterized

in that the reagent is di-n-butylamine (DBA).

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REMARKS

This Preliminary Amendment puts the claims into proper form for examination. Note that claims 1-5, 7-17, 19, and 20 have been amended; new claims 23-27 have been added; and claims 6, 18, 21, and 22 remain unchanged. Kindly calculate the filing fee based on the amended claims.

The Examiner is encouraged to telephone the undersigned attorney to discuss any matter which would expedite allowance of the present application.

Respectfully submitted,

GUNNAR SKARPING, ET AL.

By MMH/Wit

Charles I Gagnebin III Registration No. 25,467 Attorney for Applicant(s)

WEINGARTEN, SCHURGIN, GAGNEBIN

& LEBOVICI LLP

Ten Post Office Square

Boston, MA 02109

Telephone: (617) 542-2290 Telecopier: (617) 451-0313

Date: //- 38 -/

CLG/mc/262970-1 Enclosure

Red-lined claims for the Examiner's convenience:

- 1. A sampling device for analysis of a substance, which is selected from the group consisting of isocyanates, aminoisocyanates, isothiocyanates, amines and carboxylic acids and which is present in both gas and particle phase in an air flow intended to pass through the sampling device, characterized in that it comprises
- a) an adsorption device (1)—intended for the passage of the air flow and provided with a coating of a mixture of a reagent in the form of primary or secondary amines and a carboxylic acid for adsorption of and reaction with the substance in the gas phase of the air flow,
- b) a filter device (2)—intended for the passage of the air flow and provided with the mixture of reagent and carboxylic acid for adsorption of and reaction with the substance in the particle phase of the air flow, and
- c) a reagent container (4) containing the reagent,

the reagent container $\frac{(4)}{4}$ being connected to the adsorption device $\frac{(1)}{4}$ and/or the filter device $\frac{(2)}{4}$ by means of a switch device $\frac{(5)}{4}$ for conveying the reagent to these for reaction therein with the non-reacted substance.

Filed: Herewith

Group Art Unit:

2. A sampling device according to claim 1, characterized in

that the adsorption device $\frac{(1)}{}$ is tubular in the air flow

direction, the proportion of the length to the inner diameter

being superior to 5, and preferably about 10.

3. A sampling device according to claim 2, characterized in

that the inner walls of the adsorption device $\frac{(1)}{}$ are coated

with the mixture of reagent and carboxylic acid.

4. A sampling device according to any one of the preceding

claims 1, characterized in that the mixture contains one or more

different carboxylic acids, preferably formic acid, acetic acid

or propionic acid, or a combination thereof.

5. A sampling device according to any one of the preceding

claims 1, characterized in that the reagent is di-n-butylamine

(DBA).

6. A sampling device according to claim 1, characterized in

that the primary or secondary amines are volatile in unbound

form and each have a molecular weight inferior to 300.

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Filed: Herewith

Group Art Unit:

7. A sampling device according to claim 1, characterized in

that the adsorption device (1) has an upper end with an air

inlet (6) and a lower end which is connected to an upper end of

the filter device $\frac{(2)}{(2)}$, the filter device $\frac{(2)}{(2)}$ having a lower end

with an air outlet (7).

8. A sampling device according to claim 7, characterized in

that the sampling device is inverted in such manner that the air

inlet $\frac{6}{6}$ is arranged in the lower end of the filter device $\frac{2}{6}$

and that the air outlet $\frac{(7)}{}$ is arranged in the upper end of the

adsorption device (1).

9. A sampling device according to claim 1, characterized in

that it is provided with a pump or suction device (3) to provide

the passage of the air flow through the adsorption device - (1)

and the filter device (2), and that it is connected either to

the adsorption device $\frac{(1)}{(2)}$ or to the filter device $\frac{(2)}{(2)}$.

10. A sampling device according to claim 9, characterized in

that the pump or suction device $\frac{(3)}{}$ is connected to the lower

end of the filter device (2).

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Filed: Herewith

Group Art Unit:

11. A sampling device according to claim 10, characterized

in that the pump or suction device $\frac{(3)}{}$ is a vacuum tube or a

displacement pump, preferably a hose pump, diaphragm pump,

injection pump or gear-type pump.

12. A sampling device according to claim 1, characterized in

that the adsorption device (1) consists of a body packed with

particles which are coated with the mixture of reagent

carboxylic acid.

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 13. A sampling device according to claim 1, characterized in

that the filter device $\frac{(2)}{}$ is formed as a substantially flat

cylinder having a diameter which is greater than or equal to

that of the adsorption device—(1), that it contains particles

which are coated with the reagent and mixed with carboxylic

acid, and that it has an average pore diameter of 0.1-20 μm ,

preferably 0.2-2 μm , and most preferably 0.4 μm .

14. A sampling device according to claim 1, characterized in

that the reagent container (4)—is connected by means of the

switch device $\frac{(5)}{}$ to the adsorption device $\frac{(1)}{}$ or the filter

device (2)—for automatic conveyance thereto of reagent after

completed sampling and closing of the air inlet (6) and the air

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Filed: Herewith

Group Art Unit:

outlet $\frac{\langle 7 \rangle}{\langle 7 \rangle}$, respectively, for reaction with residual non-reacted

substance in the sampling device.

15. A sampling device according to claim 2, characterized in

that the reagent container (3)—is arranged as a concentric tube

round the adsorption device (2).

16. A sampling device according to claim 1, characterized in

that the adsorption device $\frac{1}{2}$ and the filter device $\frac{2}{2}$ are

enclosed in a uniform casing having a closable air inlet (6) and

outlet-(7), respectively.

17. A kit comprising a plurality of sampling devices

according to any one of the preceding claims 1 for taking

samples from a plurality of the different substances at the same

time or at different points of time, the reagent in each

sampling device being selected for the substance with which it

is to react in the air flow.

18. A method for immobilization of volatile primary and

secondary amines, preferably di-n-butylamine, on a surface,

characterized in that the volatile primary or secondary amine is

mixed with a carboxylic acid, preferably formic acid, acetic

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acid or propionic acid, and that the surface is subsequently coated with the prepared mixture.

19. A method according to claim 18, characterized in that for immobilization of volatile primary and secondary amines, preferably di-n-butylamine, on a surface, characterized in that:

the volatile primary or secondary amine is mixed with a carboxylic acid, preferably formic acid, acetic acid or propionic acid, and that the surface is subsequently coated with the prepared mixture;

one or more different primary and/or secondary amines, preferably di-n-butylamine, is/are immobilized in an adsorption device $\frac{1}{2}$ and/or a filter device $\frac{2}{1}$ in a sampling device or a kit according to any one of claims $1-\frac{17}{2}$.

20. A method for taking samples from a substance, which is selected from the group consisting of isocyanates, aminoisocyanates, isothiocyanates, amines and carboxylic acids and which is present in both gas and particle phase in an air flow, by means of a sampling device according to any one of claims 1-17, characterized in that it comprises the steps of

passing the air flow through the adsorption device $\frac{(2)}{(2)}$ and the filter device $\frac{(2)}{(2)}$ by means of the pump or suction device $\frac{(3)}{(2)}$

Filed: Herewith

Group Art Unit:

for adsorption and reaction therein of the substance in both gas

phase and particle phase,

closing the inlet $\frac{(6)}{}$ and the outlet $\frac{(7)}{}$ of the sampling

device after the passage of a predetermined volume of air, the

switch device (5) associated with the reagent container (4)

automatically securing conveyance of the reagent therein to the

adsorption device $\frac{(1)}{}$ and the filter device $\frac{(2)}{}$ for reaction

therein with the residual, non-reacted substance, and of

finally subjecting the sampling device to a qualitative

and/or quantitative analysis,

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optionally carrying out an initial activation step, in which

the mixture of the reagent and the carboxylic acid is added to

the adsorption device $\frac{(1)}{}$ and the filter device $\frac{(2)}{}$ immediately

before the initiation of the sampling.

21. A method according to claim 15, characterized in that

isocyanates, aminoisocyanates, isothiocyantes, amines and

carboxylic acids are adsorbed by and react with di-n-butylamine

in the sampling device.

22. A method according to claim 16, characterized in that an

air flow of 0.001-1 l/min, and preferably 10-100 ml/min,

passed through the sampling device.

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SAMPLING DEVICE

FIELD OF THE INVENTION

The present invention relates to a device and a method of sampling for analysis of isocyanates, amino-isocyanates, amines, isothiocyanates and carboxylic acids which are present in both gas and particle phase in an air flow.

BACKGROUND OF THE INVENTION

Polyurethane (PUR) products frequently occur in industry, in particular in manufacturing and handling polyurethane foam, elastomers, adhesives and lacquers. Polyurethane is produced by the reaction of a bifunctional isocyanate with a polyfunctional alcohol. The satisfactory technical qualities of polyurethane have resulted in a large increase of its use and application fields during the last decade. In connection with thermal decomposition of polyurethanes, however, the formation of isocyanates, aminoisocyanates and amines might occur, and extremely high contents can be found in air, e.g. when welding automobile sheet steel. Besides the known types of isocyanate, also new types of aliphatic isocyanates have been detected, in connection with e.g. heat treatment of car paint. Most of the isocyanates formed have been found to be represented by so-called low-molecular isocyanates. During short periods of time (peak exposure) particularly high isocyanate contents can be present, as is the case, for instance, when welding. Of all the dangerous substances on the limit value list, isocyanates have the lowest permissible contents. Exposure to this new type of isocyanates was previously unheard of. Isocyanates in both gas and particle phase have been detected in connection with welding, grinding and cutting of painted automobile sheet steel, and respirable particles in high contents containing isocyanates have been detected. In thermal decomposition products of painted automobile sheet steel, detection has been made

of, among other things, methyl isocyanate (MIC), ethyl isocyanate (EIC), propyl isocyanate (PIC), phenyl isocyanate (PhI), 1,6-hexamethylene diisocyanate (HDI), isophorone diisocyanate (IPDI), 2,4- and 2,6-diisocyanate toluene (TDI) and 4,4-methylene diphenyldiisocyanate (MDI).

In thermal decomposition of phenol/formaldehyde/ urea-(FFU)-plastic, isocyanic acid and methyl isocyanate are formed. FFU plastic is used, among other things, in wood glue and as a binder in mineral wool (and bakelite), which is frequently used as insulation for ovens and furnaces in industrial and domestic use. New fields of application in which exposure to isocyanates has been detected are the soldering and processing of printed circuit boards in the electronic industry, the welding, grinding and cutting of painted sheet steel in the automobile industry and the welding of lacquered copper pipes. Isocyanates have a varying degree of toxicity to the organism depending on their chemical and physical form. As a result, the hygienic limit values have been set at an extremely low level in all countries. For the exposed individual, the degree of exposure to isocyanates varies considerably in different operations during a working day and in connection with breakdowns. Thermal decomposition products from PUR constitute a special problem, since new and completely unknown isocyanates are formed, whose toxicity has not yet been analyzed in a satisfactory manner. Furthermore, the increasingly sophisticated measuring methods have revealed exposure to isocyanates in an increasing number of operations in industry.

To sum up, there is a number of operations in numerous working areas where people are daily exposed to or at risk being exposed to isocyanates at a varying degree. Considering the ominous tendency of isocyanates to cause respiratory diseases and the fact that there are some carcinogenic substances among the thermal decomposi-

tion products of polyurethane, e.g. 2,4-diamine toluene (TDA), 4,4-methylene diamine (MDA) and MOCA, it is very important to measure in a reliable, sensitive and rapid manner any presence of isocyanates, but also other decomposition products dangerous to health, in environments where there is such a risk.

Due to the high degree of reactivity of the isocyanates with other substances containing active hydrogen, the major part of the methods utilized for measuring in air flows are based on derivatisation in connection with the sampling step in order to protect the isocyanate group and allowing a selective determination of the isccyanates. A number of reagents and methods have been presented for the determination of isocyanates. However, there is only a limited amount of information about the reaction rate of isocyanates, and losses due to the presence of interfering substances has been reported, for instance, for 1-(2-methoxyphenyl)piperazine (2MP) and MAMA as derivatisation reagents for 2,4- and 2,6-TDI. A method recently developed by the present inventor has a number of advantages in comparison with the abovementioned MAMA method. This new method, which is called the DBA method due to the use of di-n-butylamine as reagent, allows the analysis of several new types of isocyanates and has been suggested as an international ISO reference method. The DBA method is based on the gathering of isocyanates in impinger bottles containing DBA in toluene and having a filter which is coupled in series and situated after the impinger bottle in the flow direction. In a sampling process, DBA solution and toluene are added to an impinger bottle. Subsequently, the sample flow is calibrated. An air flow is drawn through a tube immersed in the reagent solution, and isocyanates in the air flow react with DBA in the solution. Non-reacted gaseous isocyanates which have passed the solution are drawn through a filter which is provided with a reagent and arranged in connection with the suc-

tion device. Thus on this filter isocyanates which have not reacted with the reagent solution are bound. After completed sampling, the DBA solution with bound isocyanates is conveyed to and the filter is applied to one and the same test tube for further transport to an analysis step. Impinger bottles containing 10 ml 0.01 mole DBA in toluene have been used. Deuterium-labeled isocyanate DBA derivates are added to the samples and used as internal standards. Carbamate esters are formed by adding 2 ml $5\ M$ NaOH, 10 µl pyridine and 50 µl ethyl chloroformate to the samples. The so-called DBA method has been tested for isocyanates in connection with spray painting with two typical biuret and isocyanurate adducts, HDI, IPD, polymeric MDI, TDI and thermal decomposition products from PUR plastic. High reaction rates for the reaction of the isocyanates with DBA have been observed, and the method is not sensitive to interfering substances. Since DBA is easy to eliminate in connection with the processing of the sample, the subsequent chromatographic determination is facilitated, which allows the use of the reagent in high contents. Before the chromatographic determination, the organic phase is separated and evaporated until it is dry. The rest is dissolved in 500 μl acetonitrile, after which the solution is injected into a liquid chromatographic (LC-mass-spectrometric (MC)) system.

Other methods used for the determination of isocyanates have a number of drawbacks. Among other things, isocyanates which are present in both gas phase and particle phase in the air flow cannot be bound to the reagent in a satisfactory manner. Isocyanates which are present on and/or in particles, such as dust, will not be completely accessible to analysis, but will be polymerized to a kind of lump. Moreover, the reaction of the reagent with isocyanates is slow and negatively affected by interference from other substances present. In addition, the minimal sampling volume is about 0.5 l air, whereas the air flow which is obtained by means of a

battery-operated air pump usually amounts to about 1 l/min. Furthermore, conventional sampling devices require manual adding of solvents and reagents as well as manual dismounting to convey the reagent liquid and the filter with bound isocyanates to the final analysis test tube. Another drawback is that such a sampling device can be tampered with to obtain false results.

In view of this, there is a great demand for an improved device and an improved method for sampling isocyanates, but also other products dangerous to health, such as aminoisocyanates, amines, isothiocyanates and carboxylic acids, in a rapid, reliable, precise and tamperproof manner.

SUMMARY OF THE INVENTION

An object of the present invention is to eliminate the above-mentioned problems and provide a device and a method for improved sampling in an air flow for the analysis of isocyanates, aminoisocyanates, amines, isothiocyanates and carboxylic acids which are present in both gas and particle phase.

According to the invention, this object is achieved by means of a device and a method, respectively, of the type mentioned by way of introduction, which have the features stated in the appended claims 1 and 20, respectively. Preferred embodiments of the sampling device and the method, respectively, are defined in the dependent claims.

According to one aspect, the present invention relates to a sampling device for the analysis of substances which are present in both gas and particle phase in an air flow.

According to another aspect, the invention relates to a method for sampling in an air flow by means of the sampling device according to the present invention.

According to a further aspect, the present invention relates to a kit containing a set of a plurality of sampling devices which contain different reagents for taking

samples from different substances in an air flow, which is specified in claim 17.

According to yet another aspect, the present invention relates to a method for binding a reagent to a surface, preferably to a surface in an adsorption device 1 and a filter device 2 in the sampling device according to the present invention, which is specified in claim 18.

BRIEF DESCRIPTION OF THE DRAWINGS

- Fig. 1 schematically shows a preferred sampling device according to the present invention.
- Fig. 2 schematically shows an alternative embodiment of the sampling device according to the present invention.
- Fig. 3 schematically shows a further alternative embodiment of the sampling device according to the present invention.

DESCRIPTION OF PREFERRED EMBODIMENTS

The present invention is, among other things, based on a new method for the immobilization of reagents in the form of volatile primary and secondary amines on a surface. Since a number of such usable reagents are volatile, there is a great demand for being able to immobilize or stabilize reagents on surfaces, for instance in adsorption devices of different kind, in such a manner that the volatility of the reagent is reduced at the same time as its reactivity is maintained. This problem has been solved by the present inventor by first mixing the reagent with a carboxylic acid. The carboxylic function of the mixture then provides stability to the reagent. There is an excess of primary or secondary amine in relation to the carboxylic acid. Subsequently, the mixture is contacted with the surface on which the reagent is intended to be immobilized or applied, e.g. on the inside of tubes or on particles or spheres of different kind. Due to the surface tension, the mixture is partially adsorbed physically on the surface as a coating, and the otherwise volatile reagent is retained and can pursue its

activity. Any carboxylic acids can be used to contribute to the carboxylic acid function, e.g. both monovalent and polyvalent, saturated and unsaturated, but in a preferred embodiment use is made of formic acid (HCOOH), acetic acid (CH $_3$ COOH) or propionic acid (C $_2$ H $_5$ COOH). Combinations of one or more different carboxylic acids are also usable.

The primary or secondary amine which constitutes the reagent can be any amine which in free form is volatile and which has a molecular weight inferior to 300. Di-n-butylamine (DBA) is particularly preferred when analyzing isocyanates and aminoisocyanates. Other examples of usable amines are other dialkylamines which meet the above restriction on molecular weight.

The expression "primary or secondary amine" which is used here also comprises an amine which, in addition to the amine group, can contain one or more other functional groups which can facilitate the immobilization and/or adsorption of and reaction with the sample substance. As examples of such amines, mention can be made of alkanolamines, e.g. ethanolamines.

The substances, from which samples are to be taken by means of the method and the sampling device according to the present invention, are primarily isocyanates, aminoisocyanates and amines, but also isothiocyanates and carboxylic acids are possible. As mentioned above, these substances are frequently present in both gas and particle phase, which has previously made it more difficult to carry out a reliable analysis. Moreover, many of these compounds are volatile and so reactive that samples cannot be taken without chemical change thereof.

The sampling device according to the present invention comprises an adsorption device 1 which, in a preferred embodiment shown in Fig. 1, is substantially elongated, preferably tubular or hollow and cylindrical, the proportion of the length to the inner diameter being more than 5, preferably about 10. Such an adsorption tube,

which is also called a "denuder", can have a length of 1 cm to 1 m and an inner diameter of 0.1 mm to 1 cm. The adsorption device 1 can be made of plastic or any other low-weight material. In the preferred embodiment with a tubular adsorption device 1, the reagent is applied or immobilized on the inner walls of the tube and mixed with carboxylic acid.

When using the sampling device, sample air containing the substance which is to be analyzed is allowed to pass through the adsorption device 1, the major content of the substance in gas phase first being adsorbed on and subsequently reacting with the reagent which is immobilized on the inside of the tube walls. However, the portion of the substance which is bound on and/or in particles is passed through the adsorption device 1 together with a small portion of the substance in gas phase which has not been adsorbed.

In another embodiment, the adsorption device 1 can consist of a bed or a plate of packed particles, e.g. of glass, silicon dioxide or plastic, on which the reagent has been immobilized in the above described manner. The dimensions of the bed are not critical, but it is preferably formed as a flat cylinder.

The sampling device according to the present invention also comprises a filter device 2, which is not critical as to dimensions, but is preferably formed as a substantially flat cylinder having an inner diameter which is greater than or equal to that of the adsorption device 1. The filter device can be of any type which provides a separation of the particle phase and the gas phase in the flow and is, for instance, made of a glass or plastic material having a pore diameter of about 0.1-20 μm , preferably 0.3-0.5 μm , and most preferably about 0.4 μm . The filter device 2 is impregnated with immobilized reagent in the same way as the adsorption device 1. Substances in solid phase, i.e. that are present on or in particles, in the passing air flow are dis-

solved from the particles in the filter device 2 and react in the same way with immobilized reagent. In the case of DBA as reagent for the reaction with and binding of isocyanates, aminoisocyanates and amines, the binding reaction takes place immediately and is not affected by interfering substances in the sample.

The sampling device according to the present invention further comprises a pumping or suction device 3 which can be of any type providing the required passage of the air flow through the sampling device, but it is preferably a suction device in the form of a vacuum tube or a displacement pump, such as a hose pump, diaphragm pump, injection pump or a gear-type pump. In the preferred embodiment, this device is preferably arranged in the lower end of the sampling device, that is after the end of the filter device 2 for the discharge of the air flow. In addition, the pump or suction device 3 should not be integrated in the sampling device, but be usable more than once in contrast to a disposable sampling device. Furthermore, it should be provided with a measuring device for determining the desired amount of air that is to pass. This amount is controlled by the permissible value limit for the substance involved. The pump or suction device 3 can also be adjusted so that the passage of air is controlled in such a manner that a constant air flow is obtained during the time of sampling.

As shown in Fig. 1, in a preferred embodiment of the sampling device according to the present invention the adsorption device 1, the filter device 2 and the pump or suction device 3 are arranged in such a manner that the filter device 2 is arranged between the adsorption device 1 and the pump or suction device 3. Moreover, in this preferred embodiment the adsorption device 1 is a cylindrical adsorbent tube (denuder) comprising a reagent which has been immobilized or applied on the inside of the tube. In operation, air enters through an air inlet 6, through the adsorbent tube 1 and then through the

filter device 2 before the air flow leaves through an air outlet 7 in connection with the lower end of the filter device 2. In the most preferred embodiment, an air flow containing isocyanates, aminoisocyanates, isothiocyanates, amines and/or carboxylic acids passes through the sampling device, whose adsorbent tube 1 and filter device 2 are impregnated with di-n-butylamine (DBA). The major content of these substances in gas phase are adsorbed in and react with the reagent in the adsorption tube 1, whereas the major content of these substances in particle phase are adsorbed in and react with the reagent in the filter device 2.

However, as regards amines in the air flow, no reaction takes place with the reagent, but the amines form ion pairs with the carboxylic acids in the coating consisting of the mixture of reagent and carboxylic acids, which results in the formation of a salt.

Fig. 2 shows an alternative embodiment of the sampling device according to the present invention. The only difference in relation to the sampling device in Fig. 1 is that the adsorption device 1 and the filter device 2 are inverted, which means that as an air flow passes the major content of the substance in particle phase is first adsorbed, after which the major content of the substances in gas phase is adsorbed.

In addition, the sampling device according to the present invention comprises a reagent container 4. The reagent container 4 contains the same reagent as that immobilized in mixture with carboxylic acid in the adsorbent device 1 and the filter device 2. However, there is no carboxylic acid in the reagent container 4, and the reagent can be more or less dissolved in an organic solvent, e.g. toluene or acetonitrile, but not in alcohol. The design of the reagent container 4 is not critical, but it is preferably tubular and arranged in parallel with the adsorption tube 1. Alternatively, the reagent container 4 can be arranged concentrically with the

adsorption tube 1 and thus enclose the same. Moreover, the reagent container 4 can alternatively be connected to the filter device 2. In the preferred embodiment, the reagent container 4 is, however, connected to the tubular adsorbent device 1. When a desired air flow has passed through the sampling device according to the present invention, the air inlet 6 and the air outlet 7 are closed by means of suitable conventional closing devices. Thus a closed system is provided, in which, however, there is usually a small amount of non-adsorbed substance left in both gas phase and particle phase. To allow a complete and exact analysis of the substance which is to be analyzed, e.g. isocyanates, the reagent is let into this closed system from the reagent container 4 and reacts with the above non-reacted substance. Preferably, this takes place essentially automatically when the sampling device has been closed, but can also be carried out manually with the aid of a control means which is arranged on the outside of the sampling device. The conveyance of the reagent can, for instance, take place automatically the moment the sampling device, after sampling, is removed from its position, e.g. some kind of attachment. There is, of course, an excess of reagent in the reagent container 4 in relation to the estimated amount of nonreacted substance in the above-mentioned closed system.

The reagent container 4 can be integrated in the sampling device or detachably arranged. A switch device 5, which is situated between the reagent container 4 and the adsorption device 1 or the filter device 2, can be any conventional valve which can be opened and closed and which secures the conveyance of reagent to the adsorption device 1 and the filter device 2.

As mentioned above, the part of the sampling device which includes the adsorption device 1 and the filter device 2 can be made in one piece. Thus a spill-proof and tamperproof sampling device that is easy to handle is provided for exact measuring of the amount of a particu-

lar substance in an air flow. In addition, the sampling device can easily be kept in one's pocket, and in a manner which is advantageous in terms of security it can easily be sent on for a final analysis, e.g. by means of liquid chromatography and mass spectrometry.

If, before sampling, the sampling device is to be stored for such a long time that the stability of the reagent immobilized in the adsorption device 1 and the filter device 2 is at risk, the immobilization can instead take place immediately before the sampling by adding the mixture of reagent and carboxylic acid to the devices 1 and 2, but this must be done early enough to allow a complete coating and immobilization to take place. This so-called activation of the sampling device can be included as an optional step in the sampling method, in particular when using unstable reagents, e.g. for measuring aldehydes. Before the activation step, the mixture can be stored in a special container which is connected to the sampling device, and the addition can be carried out by means of a switch device, e.g. a valve, which can be controlled manually or more or less automatically.

In the sampling method according to the present device, the inventive sampling device, which has been manufactured according to the above-described method for immobilization of the reagent, is placed or kept at the location where the sampling of the air flow is to take place for analysis of the specific substance. The pump or suction device 3 is set at a desired flow rate according to the permissible limit value for the substance to be analyzed.

By means of the present invention, the total amount of the substance in question in the air flow can thus be quantitatively determined in a manner which was previously not possible. If desired, the amount of the substance in gas phase can be determined separately, as well as the amount of the substance in the particle phase. However,

in most cases it is above all interesting to determine at the same time the total amount of the substance in both gas and particle phase, which is achieved with the aid of the preferred embodiment of the present invention.

The sampling device according to the present invention can also be used for direct determination of the substance in question, in which case a color indicator, for instance, is brought into contact with the reacted substance in or adjacent to the sampling device.

EXAMPLE

In an experiment with an embodiment of the sampling device according to the present invention, an adsorption device (1) was used which was based on a denuder tube, whereas the filter device (2) consisted of a glass fiber filter of the type A/E (SKC, PA, USA) having a diameter of 13 mm, a thickness of 1 mm and a pore size of 0.3 μ m. The denuder tube and the filter had previously been impregnated with 100 and 50 μ l, respectively, of a reagent solution, which was prepared by adding 0.5 ml pure di-n-butylamine (DBA) and 0.5 ml concentrated acetic acid to 5 ml toluene under stirring. After the addition of this reagent solution to the denuder tube and the filter, respectively, the solvent was allowed to evaporate. The filter in the sampling device is placed in a filter holder made of teflon (Millipore Swinnex 13, Milford, MA, USA).

A reagent container containing pure DBA in toluene is connected to the denuder tube in the sampling device by means of a conventional valve. In one experiment, known amounts of isocyanates, i.e. 0.3 μ g phenylisocyanate, 0.3 μ g hexamethylene diisocyanate and 0.4 μ g toluene diisocyanate, were placed in glass tubes in front of the inlet of the sampling device. Air was passed through the sampling device by means of a conventional diaphragm pump having a flow rate of about 0.2 $1/\min$. After 2 \min , the sampling device was heated by means of a heat gun, and after a total time of sampling of 4 \min

the experiment was completed. DBA and toluene in the reagent container were passed through the valve into the denuder tube to react with non-reacted isocyanates in the denuder tube and the filter. The toluene which was added to the denuder tube and the filter dissolves the reaction product which is formed when the isocyanates have reacted with DBA, and therefore this reaction product is completely dissolved in the sampling device, i.e. it is not left immobilized on the inner walls of the denuder tube or on the surface of the filter. Subsequently, a predetermined amount of an internal standard in the form of deuterium-labeled isocyanates is added to the sampling device, whose inlet and outlet are then closed before transporting the sampling device to a laboratory for analysis.

Before the laboratory analysis, the sampling device was opened, and the DBA solution which was present in the same and contained the above-mentioned reaction product was conveyed to another test tube. Subsequently, the toluene was eliminated by evaporation, after which 0.5 ml acetonitrile was added. After this, the samples were ready for analysis by liquid chromatography (LC) in connection with mass spectrometry (MS). The separation of the different isocyanate reaction products was carried out by means of LC technique and MS detection. The mass spectrometer was connected in series to an LC system. Use was made of a column of Hypersil C18 type.

The isocyanates were detected by monitoring $[M+1]^+$ ions for the DBA derivatives. Calibration plots were obtained from the proportions of the surfaces for the internal standard to those of the samples, and from which plots the amount of isocyanate in the sample was determined. The detection limits are about 0.2 μ g per isocyanate and sample.

In the performed experiment, it was found that the isocyanates gathered in the sampling device at a yield of 100 ± 10 %.

CLAIMS

- 1. A sampling device for analysis of a substance, which is selected from the group consisting of isocyanates, aminoisocyanates, isothiocyanates, amines and carboxylic acids and which is present in both gas and particle phase in an air flow intended to pass through the sampling device, characterized in that it comprises
- a) an adsorption device (1) intended for the passage of the air flow and provided with a coating of a mixture of a reagent in the form of primary or secondary amines and a carboxylic acid for adsorption of and reaction with the substance in the gas phase of the air flow,
- b) a filter device (2) intended for the passage of the air flow and provided with the mixture of reagent and carboxylic acid for adsorption of and reaction with the substance in the particle phase of the air flow, and
- c) a reagent container (4) containing the reagent,

the reagent container (4) being connected to the adsorption device (1) and/or the filter device (2) by means of a switch device (5) for conveying the reagent to these for reaction therein with the non-reacted substance.

- 2. A sampling device according to claim 1, c h a r a c t e r i z e d in that the adsorption device (1) is tubular in the air flow direction, the proportion of the length to the inner diameter being superior to 5, and preferably about 10.
- 3. A sampling device according to claim 2, c h a r a c t e r i z e d in that the inner walls of the adsorption device (1) are coated with the mixture of reagent and carboxylic acid.
- 4. A sampling device according to any one of the preceding claims, characterized in that the mixture contains one or more different carboxylic acids, preferably formic acid, acetic acid or propionic acid, or a combination thereof.
- 5. A sampling device according to any one of the preceding claims, characterized in that the reagent is di-n-butylamine (DBA).
- 6. A sampling device according to claim 1, characterized in that the primary or secondary amines are volatile in unbound form and each have a molecular weight inferior to 300.
- 7. A sampling device according to claim 1, c h a r a c t e r i z e d in that the adsorption device (1) has en upper end with an air inlet (6) and a lower end which is connected to an upper end of the filter device (2), the filter device (2) having a lower end with an air outlet (7).
- 8. A sampling device according to claim 7, c h a r a c t e r i z e d in that the sampling device is inverted in such manner that the air inlet (6) is arranged in the lower end of the filter device (2) and that the air outlet (7) is arranged in the upper end of the adsorption device (1).

- 9. A sampling device according to claim 1, c h a r a c t e r i z e d in that it is provided with a pump or suction device (3) to provide the passage of the air flow through the adsorption device (1) and the filter device (2), and that it is connected either to the adsorption device (1) or to the filter device (2).
- 10. A sampling device according to claim 9, c h a r a c t e r i z e d in that the pump or suction device (3) is connected to the lower end of the filter device (2).
- 11. A sampling device according to claim 10, c h a r a c t e r i z e d in that the pump or suction device (3) is a vacuum tube or a displacement pump, preferably a hose pump, diaphragm pump, injection pump or gear-type pump.
- 12. A sampling device according to claim 1, c h a r a c t e r i z e d in that the adsorption device (1) consists of a body packed with particles which are coated with the mixture of reagent and carboxylic acid.
- 13. A sampling device according to claim 1, c h a r a c t e r i z e d in that the filter device (2) is formed as a substantially flat cylinder having a diameter which is greater than or equal to that of the adsorption device (1), that it contains particles which are coated with the reagent and mixed with carboxylic acid, and that it has an average pore diameter of 0.1-20 μ m, preferably 0.2-2 μ m, and most preferably 0.4 μ m.
- 14. A sampling device according to claim 1, c h a r a c t e r i z e d in that the reagent container (4) is connected by means of the switch device (5) to the adsorption device (1) or the filter device (2) for automatic conveyance thereto of reagent after completed sampling and closing of the air inlet (6) and the air outlet (7), respectively, for reaction with residual non-reacted substance in the sampling device.

- 15. A sampling device according to claim 2, c h a r a c t e r i z e d in that the reagent container (3) is arranged as a concentric tube round the adsorption device (2).
- 16. A sampling device according to claim 1, c h a r a c t e r i z e d in that the adsorption device 1 and the filter device 2 are enclosed in a uniform casing having a closable air inlet (6) and outlet (7), respectively.
- 17. A kit comprising a plurality of sampling devices according to any one of the preceding claims for taking samples from a plurality of the different substances at the same time or at different points of time, the reagent in each sampling device being selected for the substance with which it is to react in the air flow.
- 18. A method for immobilization of volatile primary and secondary amines, preferably di-n-butylamine, on a surface, characterized in that the volatile primary or secondary amine is mixed with a carboxylic acid, preferably formic acid, acetic acid or propionic acid, and that the surface is subsequently coated with the prepared mixture.
- 19. A method according to claim 18, c h a r a c t e r i z e d in that one or more different primary and/or secondary amines, preferably di-n-butylamine, is/are immobilized in an adsorption device (1) and/or a filter device (2) in a sampling device or a kit according to any one of claims 1-17.
- 20. A method for taking samples from a substance, which is selected from the group consisting of isocyanates, aminoisocyanates, isothiocyanates, amines and carboxylic acids and which is present in both gas and particle phase in an air flow, by means of a sampling device according to any one of claims 1-17, characterized in that it comprises the steps of

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passing the air flow through the adsorption device (1) and the filter device (2) by means of the pump or suction device (3) for adsorption and reaction therein of the substance in both gas phase and particle phase,

closing the inlet (6) and the outlet (7) of the sampling device after the passage of a predetermined volume of air, the switch device (5) associated with the reagent container (4) automatically securing conveyance of the reagent therein to the adsorption device (1) and the filter device (2) for reaction therein with the residual, non-reacted substance, and of

finally subjecting the sampling device to a qualitative and/or quantitative analysis,

optionally carrying out an initial activation step, in which the mixture of the reagent and the carboxylic acid is added to the adsorption device (1) and the filter device (2) immediately before the initiation of the sampling.

- 21. A method according to claim 15, characterized in that isocyanates, aminoisocyanates, isothiocyantes, amines and carboxylic acids are adsorbed by and react with di-n-butylamine in the sampling device.
- 22. A method according to claim 16, characterized in that an air flow of 0.001-1 l/min, and preferably 10-100 ml/min, is passed through the sampling device.

DECLARATION AND POWER OF ATTORNEY

As a below-named inventor, I hereby declare that:

(Application Number)

My residence, post office address and citizenship are as stated below next to my name;

I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention entitled:

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weden Country)	3 June 1999 (Day/Month/Year)	[X] Yes	[] No
Country)	(Day/Month/Year)	[] Yes	[] 0M
Country)	(Day/Month/Year)	[] Yes	[] No
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(Filing Date)

Express Mail Number

EV 009950135 US

Attorney

Docket No.: AWA-043XX

I hereby claim the benefit under Title 35 USC §120 of any United States application(s) listed below and insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States application in the manner provided by the first paragraph of Title 35 USC §112, I acknowledge the duty to disclose material information as defined in Title 37, Code of Federal Regulations, §1.56(a) which occurred between the filing date of the prior application and the national or PCT international filing date of this application:

(Application No.)	(Filing date)	(Patented/pending/abandoned)
(Application No.)	(Filing date)	(Patented/pending/abandoned)
(Application No.)	(Filing date)	(Patented/pending/abandoned)

POWER OF ATTORNEY: As a named inventor, I hereby appoint the following attorney(s) to prosecute this application and transact all business connected therewith in the Patent and Trademark Office, and to file with the USRO any International Application based thereon.

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I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

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Post Office Address	City	State or Country Zip Code		
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	date in permanent ink.)	Date signed:		
	Marianne Dalene City of Residence Lund Post Office Address Stenåldersvägen 17	Marianne Dalene City of Residence State or Country Lund Sweden Post Office Address City		

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Pull Name of Joint Inventor:				
City of Residence	State or Country	Country of Citizenship		
Post Office Address	City	State or Country Zip Code		
Signature: (Please sign and	Date signed:			

Full Name of Joint Inventor:				
City of Residence	State or Country	Country of Citizenship		
Post Office Address	City	State or Country Zip Code		
Signature: (Please sign and	date in permanent ink.)	Date signed:		

Full Name of Joint Inventor:			
City of Residence	State or Country	Country of Citizenship	
Post Office Address	City	State or Country Zip Code	
Signature: (Please sign and	gnature: (Please sign and date in permanent ink.)		